## Finding Polar Covalent Bonds Name\_\_\_\_\_

For each of the molecules below do the following:

- a) determine the electronegativities of each atom in the molecule
- b) circle all polar covalent bonds
- c) label atoms at either end of each polar covalent bond with the appropriate symbol
  - $\partial$  for partially negative and  $\partial$  + for partially positive

## 1-pentanol

## 1-3 propanediol

## glycerol

If you had a beaker of each substance above, which one would have the highest boiling point? Why?